

F. Nunziata

Why we do care Natural media Mixture

Dielectric constant

Propagation issues Average value Uniform plane wave Special media

Cells Polarization

Biological tissues Debye model Dispersion in biological tissues non-Debye models Cole-Cole model

Microtubules

EM wave propagation in complex media

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- Polarization
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 - Electroorientation



Frequency behavior of material media

Frequency response

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The characterization of the frequency response of material media is of paramount importance to fully understand the behavior of EM within the material media. This can be done by describing the frequency behavior of the:

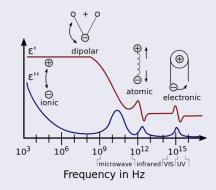
- Complex permittivity $\epsilon(\omega)$.
- Complex permeability $\mu(\omega)$.
- Complex conductivity $\sigma(\omega)$.

A material that calls for a non-flat frequency behavior of at least one of its electrical parameters is termed as dispersive.



Dispersion

Dispersion



Dispersion depends on the mechanisms, occurring at micro-scale, in the dielectric medium, e.g.; ionic and dipolar relaxation, and atomic and electronic resonances at higher energies

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They can be roughly classified as:

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- Homogeneous. Material media whose properties do not depend on space, e.g., pure water, ice, etc.
- Heterogeneous. Material media whose properties are space variant, e.g., sea ice, dry soil, wet soil, etc.

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Remote sensing - sea ice

This photomicrograph shows a thin section of sea ice containing brine channels. — Credit: Weeks and Assur, 1969

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Cell and organelles



Natural media

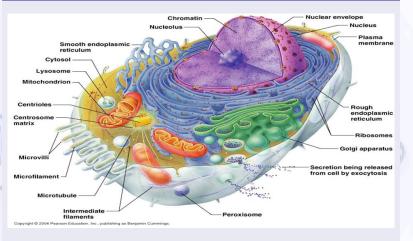
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Living cells are heterogeneous mixtures

They are of high complexity. In addition to the different ions and biomolecules, the membranous organelles inside the cells add more interfaces within the cell.

Frequency response

Dielectric properties of cells are frequency-dependent The relative permittivity is always high at very low frequencies, and by increasing frequencies, the values tend to suffer from stepwise decrements or dielectric dispersions that occur at specific range of frequencies.



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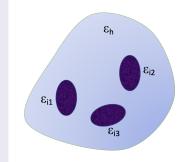
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Heterogeneous substances



The heterogeneous medium consists of an host material, whose dielectric constant is ϵ_h , in which there is a random concentration of ellipsoidal particle with a dielectric constant ϵ_i .



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Dielectric constant of the mixture

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The dielectric constant of the mixture is given by:

The average value of the dielectric constant

•
 •
 e_a(
 p) = ⟨
 e_m(**r**;
 p)⟩ is the average (or effective) value of the dielectric constant of the mixture.

 $\epsilon_m(\mathbf{r}; \hat{\mathbf{p}}) = \epsilon_a(\hat{\mathbf{p}}) + \epsilon_f(\mathbf{r}; \hat{\mathbf{p}})$

- It is not a function of the position.
- It may depend on the polarization p̂ of the incident wave. This happens when the ellipsoids that characterize the inclusions have a preferred orientation. If the latter are randomly oriented there is no polarization selectivity.

It affects the propagation constant of the EM wave

(1)



Dielectric constant of the mixture

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The fluctuating component of the dielectric constant

- ϵ_f(**r**; **p̂**) is the fluctuating component that mainly accounts for the deviations wrt the average value of the dielectric constant.
 - It depends on the spatial coordinates.
 - It depends on the polarization of the incident wave.
 - Its average value is equal to zero.

Its statistical spatial distribution is linked to volumetric scattering



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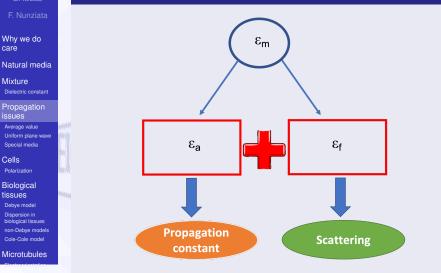
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Dielectric constant of the mixture

Dielectric constant -> wave propagation





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Dielectric constant

Average value of the dielectric constant

From now on, the average value of the dielectric constant *ε_a*(·) is termed simply as dielectric constant and is written as *ε*.

$$\epsilon = \epsilon_a = \epsilon' - j\epsilon'' \tag{2}$$

• ϵ' is typically termed as dielectric constant.

- $\epsilon'' = \frac{\sigma}{\omega \epsilon_0}$ is the dielectric loss factor.
- The conductivity of a material is a measure of the ability of its charge to be transported throughout its volume by an applied electric field.
- Its permittivity is a measure of the ability of its dipoles to rotate or its charge to be stored by an applied external field 15/89

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EM wave propagation

Complex permittivity

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Permittivity characterizes the electrical properties of materials:

- The real part gives the contrast with respect to vacuum.
- The imaginary part gives the electromagnetic loss of the material.
- Loss tangent is defined as the imaginary part divided by the real part of permittivity.

$$tan\delta = \frac{\epsilon''}{\epsilon'} = \frac{\sigma}{\omega\epsilon}$$
(3)



EM wave propagation

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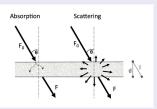
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Total em loss in a medium consists of:

- Absorption loss (electromagnetic power transformed into other forms of energy, such as heat).
- Scattering loss (energy is caused to travel in directions other than that of incident radiation).
- Penetration depth provides an approximate value to the maximum depth of the medium that contributes to the signal backscattering.





Displacement and conduction currents

From the Maxwell-Ampere law:

 $\nabla \times \mathbf{H} = \mathbf{J}_{o} + \mathbf{J}_{c} + j\omega\epsilon\mathbf{E}$ $\mathbf{J}_{o} + \sigma_{s}\mathbf{E} + j\omega\left(\epsilon' - j\epsilon''\right)\mathbf{E}$ $\mathbf{J}_{o} + \left(\sigma_{s} + \omega\epsilon''\right)\mathbf{E} + j\omega\epsilon'\mathbf{E}$ $\mathbf{J}_{o} + \sigma_{l}\mathbf{E} + j\omega\epsilon'\mathbf{E}$

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- σ_s is the static field conductivity.
- σ_l is the equivalent conductivity that is equal to $\sigma_s + \omega \epsilon^{''} = \sigma_s + \sigma_a$, with σ_a termed as alternating conductivity.

(4)



Dielectric constant of the mixture

Dielectric constant -> wave propagation

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0.5 0	GHz	2 GHz		5 GHz	
ϵ'_r	σ	ϵ_r'	σ	ϵ_r'	σ
44.91	0.728	38.56	1.265	35.77	3.06
48.62	0.704	43.52	1.335	39.61	3.574
5.54	0.042	5.32	0.085	5.02	0.242
57.32	0.843	54.16	1.508	50.13	4.24
12.94	0.10	11.65	0.31	10.04	0.962
41	0.473	36.73	1.001	33.44	2.858
63.25	1.383	59.02	2.186	53.95	5.395
	 ε'r 44.91 48.62 5.54 57.32 12.94 41 	44.91 0.728 48.62 0.704 5.54 0.042 57.32 0.843 12.94 0.10 41 0.473	ϵ_r' σ ϵ_r' 44.91 0.728 38.56 48.62 0.704 43.52 5.54 0.042 5.32 57.32 0.843 54.16 12.94 0.10 11.65 41 0.473 36.73	e'_r σ e'_r σ 44.91 0.728 38.56 1.265 48.62 0.704 43.52 1.335 5.54 0.042 5.32 0.085 57.32 0.843 54.16 1.508 12.94 0.10 11.65 0.31 41 0.473 36.73 1.001	ϵ_r' σ ϵ_r' σ ϵ_r' 44.91 0.728 38.56 1.265 35.77 48.62 0.704 43.52 1.335 39.61 5.54 0.042 5.32 0.085 5.02 57.32 0.843 54.16 1.508 50.13 12.94 0.10 11.65 0.31 10.04 41 0.473 36.73 1.001 33.44

Frequency

non-Debye models Cole-Cole model Microtubules

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Maxwell-Ampere law

Equivalent conductivity

The equivalent conductivity (σ_l) consists of the static portion σ_s and the alternating one σ_a caused by the rotation of the dipoles as they attempt to align with the applied field when its polarity is alternating.

The phenomenon (rotation of dipoles) that contributes the alternating conductivity is referred as dielectric hysteresis and generates heat. The heat generated by this radio frequency process is used for:

Industrial heating processes (e.g., microwave cooking).

Selective heating of human tissue for tumor treatment.
 Selective heating of certain compounds in materials.

Note that σ_l is frequency dependent. From now on, σ_l is written as σ .

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A uniform plane wave propagating along the \hat{z} direction

The propagation vector is given by:

$$\mathbf{S} = \mathbf{a} + j\mathbf{k} = S\hat{z} = (\alpha + j\beta)\,\hat{z} \tag{5}$$

A generic transverse component of the wave in the time domain is given by:

$$e(z,t) = E_o e^{-\alpha z} \cos(\omega t - \beta z)$$
(6)

To evaluate α and β

$$\gamma^{2} = \boldsymbol{S} \cdot \boldsymbol{S} = (\alpha + j\beta) (\alpha + j\beta) = -\omega^{2} \mu \left(\epsilon' - j \frac{\sigma}{\omega} \right) \quad (7)$$



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By equating separately real and imaginary parts one obtains (note that, without any ambiguity, hereinafter ϵ = ϵ'):

$$\beta^2 - \alpha^2 = \omega^2 \epsilon \mu \tag{8}$$

$$2\alpha\beta = \sigma\omega\mu \tag{9}$$

Solving for β eq.(9) one obtains $\beta = \frac{\sigma \omega \mu}{2\alpha}$ that can be inserted in (8) to obtain:

$$\alpha^2 + \omega^2 \mu \epsilon \alpha^2 - \frac{\omega^2 \mu^2 \sigma^2}{4} = 0 \tag{10}$$

since $k = \omega \sqrt{\epsilon \mu} = k_o n$ the following solution is obtained:

$$lpha^2=rac{-k^2\pm\sqrt{k^4+\omega^2\mu^2\sigma^2}}{2}$$

(11)



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Since the argument of the square root in eq.(11) is larger than k², to guarantee that α ≥ 0, the positive sign is selected:

 k^2

$$\alpha^{2} = \frac{-k^{2} + \sqrt{k^{4} + \omega^{2} \mu^{2} \sigma^{2}}}{2}$$
(12)

 $\omega^2 \mu^2 \sigma^2$

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which can be written as:

 α^2

(13)



The propagation vector becomes:

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Cole-Cole model

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$$\alpha = \frac{k}{\sqrt{2}} \sqrt{\sqrt{1 + \frac{\sigma^2}{\omega^2 \epsilon^2}} - 1}$$
(14)
$$\beta = \frac{k}{\sqrt{2}} \sqrt{\sqrt{1 + \frac{\sigma^2}{\omega^2 \epsilon^2}} + 1}$$
(15)

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The metric $R = \frac{\sigma}{\omega \epsilon}$ is the ratio between the norms of the conduction and the displacement currents.



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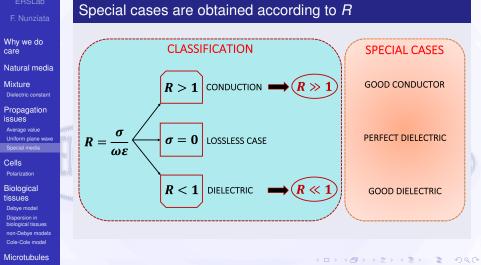
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Special cases





Special cases - Lossless

Lossless - pure dielectric

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$\begin{aligned} \alpha &= 0 \\ \beta &= k = k_o n = \omega \sqrt{\mu \epsilon} \end{aligned}$ (16) (17)

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Good dielectric: $R \ll 1$

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$$=\frac{k}{\sqrt{2}}\sqrt{\sqrt{1+R^{2}}-1}\approx\frac{k}{\sqrt{2}}\sqrt{1+\frac{R^{2}}{2}-1}=\frac{kR}{2}=\frac{\sigma}{2}\sqrt{\frac{\mu}{\epsilon}}$$
(18)

$$\beta \approx \frac{k}{\sqrt{2}}\sqrt{1 + \frac{R^2}{2} + 1} = k\sqrt{1 + \frac{R^2}{4}} \approx k\left(1 + \frac{R^2}{8}\right)$$
 (19)

 $\frac{\alpha}{\beta} \ll 1$

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Path Loss in a Wireless Body Area Network (WBAN)

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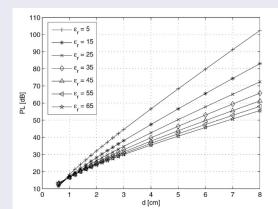
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A WBAN consists of a wireless network with devices placed close to, attached on, or implanted into the human body. Wireless communication within human body experiences loss in the form of attenuation and absorption.



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Good conductor: $R \gg 1$

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 α and β

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$\alpha \approx \beta \approx k \sqrt{\frac{R}{2}} = \sqrt{\frac{\sigma \omega \mu}{2}} = \frac{1}{\delta}$ (20)

The propagation constant becomes:

 δ is known as penetration length, i.e., the distance where the modulus of the field reaches the value $e^{-\alpha z} = e^{-\frac{z}{\delta}} = e^{-1}$.

 $\gamma = \frac{1+j}{j}$

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Good conductor - Penetration length





Good conductor: $R \gg 1$

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Remarks for a good conductor

- **Neither** α nor β depend on ϵ' .
- Eq.(20) can be equivalently rewritten as: α ≈ β ≈= √πfµσ. Since the conductivity of most materials changes very slowly with frequency, this expression indicates that α (and β) increases approximately in proportion to the square root of frequency for good conductors.
- *R* is frequency dependent, hence conductors can be "good conductors" according to the frequency.



Biological tissues

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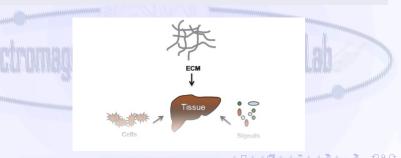
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They consist of cells, the extracellular matrix (ECM), and the signaling systems that are encoded by genes in the nuclei of the cells and then activated through cues from the ECM or other cell.





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A fundamental brick in the wall

The smallest unit that can live on its own and that makes up all living organisms and the tissues of the body. A cell has three main parts: the cell membrane, the nucleus, and the cytoplasm. The cell membrane surrounds the cell and controls the substances that go into and out of the cell.

They are composite structures

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Living cells and tissues are composed of different types of molecules, ranging from the simple free moving ions and polar water molecules, to the more complex biomolecules such as carbohydrates, proteins, DNA and lipids



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Cell polarization

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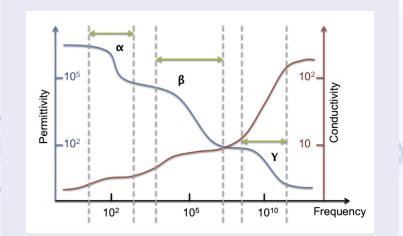
They can get polarized

Exposing cells or tissues to an external electric field affects the distribution of charges and other molecules in them, such that the ions tend to move over distances (thus, acting as conductors), while other molecules reorient themselves in space and get polarized. This makes cells a dielectric substance that has the ability to get polarized



Biological cells

Complex permittivity spectrum of biological cells



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A deeper understanding of cell dispersive behavior

Cells are modeled as highly conductive cytoplasmic spheres having free ions that are surrounded by insulating non-conductive cell membranes.

- Cell membranes separate the internal compartment from the external media which is also conductive.
- The highly negative charge of cell membranes is normally neutralized by an adsorbed cloud of counter ions forming an electric double layer



Electrical polarization

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Electrical cell polarization



Electric double layer around the cells membrane in the absence of applied electric field

Polarization of the electric double laver around the cells membrane

- Negatively charged Glycocalyx
- Positively charged ions





Random distribution of ions inside the cell and in the extracellular media



Interfacial polarization under the effect of applied electric field





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Electrical cell polarization

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Main mechanisms on a micro-scale

The Complex permittivity spectrum of biological cells shows the distinct dispersion phases at their respective frequency, permittivity and conductivity ranges.

- α-dispersions are generally associated with the diffusion processes of ionic species or electronic double-layer polarization.
- β-dispersions are due to interfacial polarization across the cellular plasma membranes and their interactions with the extra and intra-cellular electrolytes.
- γ-dispersions are caused by the aqueous content of the biological species and the presence of small molecules.



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There are 4 basic types of tissue: connective tissue, epithelial tissue, muscle tissue, and nervous tissue.

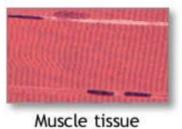
- Connective tissue supports other tissues and binds them together (bone, blood, and lymph tissues).
- Epithelial tissue provides a covering (skin, the linings of the various passages inside the body).
- Muscle tissue includes striated (also called voluntary) muscles that move the skeleton, and smooth muscle, such as the muscles that surround the stomach.
- Nerve tissue is made up of nerve cells (neurons) and is used to carry "messages" to and from various parts of the body.

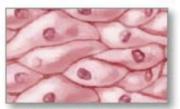


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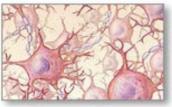


Connective tissue





Epithelial tissue



Nervous tissue



Electrical properties of biological tissues

The electrical properties of biological tissues and cell suspensions have been of interest for over a century for many reasons. They determine the pathways of current flow through the body and, thus, are very important in the analysis of a wide range of biomedical applications:

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- the functional electrical stimulation;
- the diagnosis and treatment of various physiological conditions with weak electric currents;
- radio-frequency hyperthermia;
- electrocardiography;
- body composition.



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To analyze the response of a tissue to electric stimulation, we need data on the specific conductivity and relative permittivity of the tissues or organs.

Microscopic approach

It is complicated by the variety of cell shapes and their distribution inside the tissue as well as the different properties of the extracellular media.



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Macroscopic approach

The material is described as having a permittivity and a conductivity. Even on a macroscopic level, the electrical properties are complicated. They can depend on:

- The tissue orientation relative to the applied field (directional anisotropy).
- The frequency of the applied field (the tissue is neither a perfect dielectric nor a perfect conductor).
- They can be time- and space-dependent (e.g., changes in tissue conductivity during electropermeabilization).



Dielectric properties of some tissues

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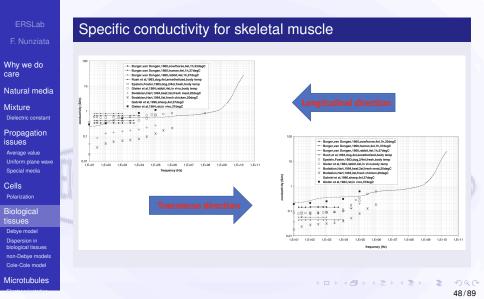
Large differences exist in electric properties of biological materials that arise, to a large extent, from the fluid content of the material.

- Relatively good conductors: blood and brain.
- Relatively poor conductors: lungs, skin, fat, and bone.
- Intermediate conductors: liver, spleen, and muscle.

Data on specific conductivity and relative permittivity of biological tissues are mostly available only at frequencies above 100 Hz. For most tissues, with the exception of anistrotropic tissues, the dielectric properties are almost no frequency-dependent between 100 Hz and 100 kHz.



Biological (anisotropic) tissues





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Biological tissues actually display some characteristics of both insulators and conductors because they contain dipoles as well as charges that can move, but in a restricted manner.

Heterogeneous materials

For materials that are heterogeneous in structure:

- Charges may become trapped at interfaces between different media.
- In addition to the conventional polarization, as positive and negative ions move in opposite directions under the applied field, internal charge separations can then result within the material, producing an effective internal polarization that acts like a very large dipole.



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Dispersion

For most materials the relative permittivity varies with the frequency of the applied signal, i.e.; σ and ϵ_r are frequency-dependent. Such a variation is called dispersion.

 Biological tissues exhibit several different dispersions over a wide range of frequencies.

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Electrical polarization

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Electrical polarization

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The most important phenomenon associated with a dielectric material is its polarization, which consists of the change of the distribution of its molecular and atomic charges when it is subjected to the action of an electric field.

When an electric field is applied to a dielectric it produces a very small electric current called dielectric loss, and its constituent particles, ions or molecules suffer small dislocations or rearrangements, thus altering their equilibrium positions.



Debye model

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Relaxation time

These molecular parts do not leave nor reach their state of equilibrium instantaneously: a variable amount of time is necessary for this change of positions to take place.

This time lapse necessary for the material to respond to the electric field applied is called relaxation time.

Debye relaxation model

In 1929, Debye conceived a simple model for the relaxation process in which he supposed a unique relaxation time for all molecules.

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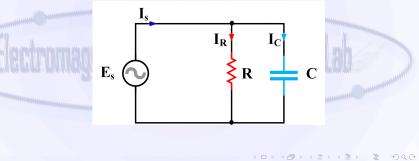
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The Debye model admits a simple circuit interpretation. We can represent the tendencies to store energy and to dissipate power by using a circuit model that consists of the parallel combination of the capacitor and conductor.





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Let's consider a sample of material whose cross-section is *A* and whose thickness is *d*.

The ability of the material to store energy can be represented by a capacitor with capacitance:

$$C = \epsilon \frac{A}{d}$$

(22)

(23)

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The ability of the material to dissipate power can be represented by a conductor with conductance:

$$G = \sigma \frac{A}{d}$$

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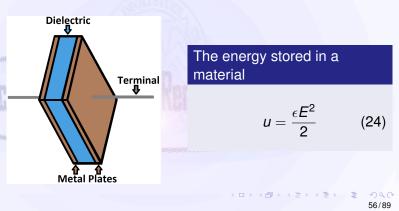
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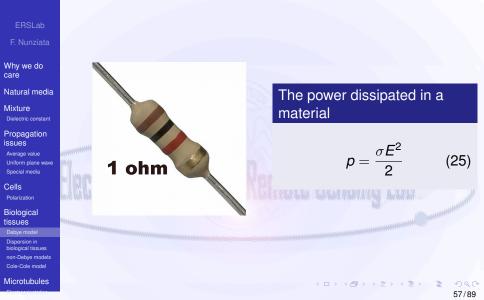
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The permittivity characterizes the material's ability to trap or store charge or to rotate molecular dipoles, whereas the conductivity describes its ability to transport charge.









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When the circuit is subjected to an alternating voltage $V(t) = V_0 cos(\omega t)$

A conduction current will flow:

$$I_c = GV \tag{26}$$

The charge Q = CV on the capacitor plates changes with the frequency f giving rise to a displacement current: $I_{d} = \frac{dQ}{dt} = -\omega CV_{o} sin(\omega t)$

The total current is the sum of the conduction and displacement currents, which are 90 degrees apart in phase:

$$I = GV + \frac{dQ}{dt} = -\omega CV_o sin(\omega t) + GV$$
 (28)

(27)



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Eq.(28) in the phasor domain reads as follows:

$$I = j\omega\epsilon \frac{A}{d}V + \sigma \frac{A}{d}V = (\sigma + j\omega\epsilon)\frac{A}{d}V$$
(29)

By factorizing
$$j\omega\epsilon_o$$
:

• with $\epsilon' = \epsilon_r$ and $\epsilon'' = \frac{\sigma}{\omega \epsilon_r}$

$$I = j\omega\epsilon_o \left(\epsilon_r - j\frac{\sigma}{\omega\epsilon_o}\right) \frac{A}{d}V$$
(30)

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The quantity in the brackets is the complex-valued relative permittivity:

$$\epsilon = \epsilon' - j\epsilon''$$

(31)

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 For a Debye-type response, which corresponds to parallel RC elements, dispersion can be represented as follows

$$\epsilon(\omega) = \epsilon_{\infty} + \frac{\epsilon_{s} - \epsilon_{\infty}}{1 - j\omega\tau} + \frac{\sigma}{j\omega\epsilon_{o}}$$
(32)

• with $\tau = \frac{1}{RC}$ being the relaxation time, ϵ_{∞} and ϵ_s represent the relative permittivity at frequencies well above and well below the dispersion.

The Debye model works fine for materials that exhibit a single relaxation time. However, this is rarely the case and, in particular, this situation does not apply for biological tissues.



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Dielectric dispersion in tissues

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The dispersion behavior

Biological tissues exhibit several different dispersion over a wide range of frequencies.

 Dispersion can be understood in terms of the orientation of the dipoles and the motion of the charge carriers.

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Limitations of Debye model

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According to Debye relaxation process:

- All dipoles in the system relax with the same relaxation time (which is called a single relaxation time approximation).
 - The system response function is purely exponential.
 - Debye relaxation appears usually in liquids or in the case of point-defects of almost perfect crystals.

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Limitations of Debye model

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Real materials show non-Debye relaxation

This fact is due to the occurrence of

- Multiple interaction processes.
- To the presence of more than one molecular conformational state or type of polar molecule.
- To polarization processes whose kinetics are not first order.
- To the presence of complex intermolecular interactions.



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Dispersion regions

- At relatively low frequencies, the permittivity is relatively high and the conductivity is relatively low. This is due to the fact that it is relatively easy for the dipoles to orient in response to the change in the applied field, whereas the charge carriers travel larger distances over which a greater opportunity exists for trapping at an interface.
- As the frequency increases, the dipoles are less able to follow the changes in the applied field, and the corresponding polarization disappears. In contrast, the charge carriers sample shorter distances during each half-cycle and are less likely to be trapped.
- As frequency increases, the permittivity decreases and, because trapping becomes less important, the conductivity increases.



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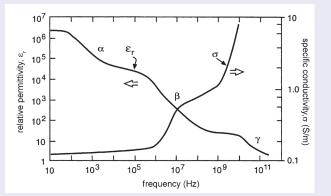
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Biological tissues exhibit several different dispersion over a wide range of frequencies.



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Dielectric response arising from multiple first-order processes

The dielectric response consists of multiple Debye terms, one for each relaxation time of the system:

$$\epsilon(\omega) = \epsilon_{\infty} + \frac{\Delta\epsilon_1}{1 - j\omega\tau_1} + \frac{\Delta\epsilon_2}{1 - j\omega\tau_2} + \dots$$
 (33)

• with $\Delta \epsilon_n$ is the limit of the dispersion characterized by time constant τ_n represent.



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Dielectric response arising from multiple first-order processes

- If the relaxation times are well separated such that τ₁ ≪ τ₂ ≪ ..., a plot of the dielectric properties as a function of frequency well exhibit clearly resolved dispersion regions.
- If the relaxation times are not well separated, the material will exhibit a broad dispersion encompassing all the relaxation times and the dispersion regions mentioned above become, in the limit, part of a continuous distribution of relaxation times.



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Dielectric response arising from multiple first-order processes

$$\epsilon(\omega) = \epsilon_{\infty} + (\epsilon_{s} - \epsilon_{\infty}) \int_{0}^{\infty} \frac{\rho(\tau)}{1 - j\omega\tau}$$
(34)

• where $\rho(\tau)$ is a normalized distribution function:

$$\int_0^\infty \rho(\tau) = 1 \tag{35}$$

According to eq.(34), all the dielectric dispersion data can be represented once an appropriate distribution function is provided and known



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Different models can be obtained by a suitable choice of $\rho(\tau)$

- Cole-Cole.
- Davidson-Cole.
- Havriliak and Negami.
- etc.

Cole-Cole model

$$\epsilon(\omega) = \epsilon_{\infty} + \frac{\epsilon_{s} - \epsilon_{\infty}}{1 - (j\omega\tau)^{\alpha}} + \frac{\sigma}{j\omega\epsilon_{o}}$$
(36)



Dielectric dispersion of water in tissues

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- ϵ_{∞} is the permittivity at field frequencies such that $\omega \tau \gg 1$ (typical values are 2.5 for bones and 3.5 for all the other tissues).
- ϵ_s is the static permittivity.
 - α is the distribution parameter bounded between 0 and 1

 σ is the ionic conductivity.

 ϵ_o is the permittivity of free space (i.e., 8.854 \times 10⁻¹²F/m).

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Dielectric dispersion of water in tissues

Cole-Cole and Debye model

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The Cole-Cole model reduces to the Debye model when $\alpha = 1$.

In circuit terms, the Cole-Cole model generalizes the Debye circuit model by replacing the capacitor with a "Constant Phase Element" (CPE) with a complex-valued impedance given by

$$Z_{CPE} = A(j\omega)^{-n} \tag{37}$$

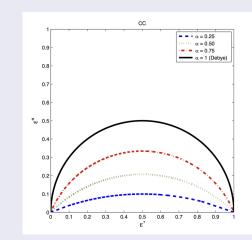
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- with A being a parameter and $n = \alpha$.
- This CPE impedance reduces to a simple resistance for n = 0 and to a capacitive reactance for n = 1.



Dielectric dispersion in tissues

Cole-Cole model at variance of α



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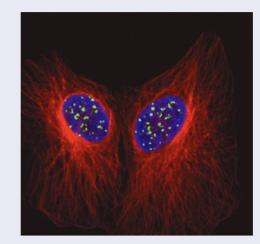
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Microtubules - see the red filaments



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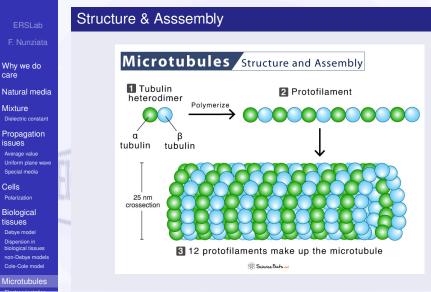
Microtubules

Microtubules are major components of the cytoskeleton. They are rigid hollow rods approximately 25 nm in diameter that undergo continual assembly and disassembly within the cell. They play a key role:

- To determine cell shape.
- To help prepare the cell for cell division and migration.
- To act as a railway track on which motor proteins transport materials within the cell.
- To the separation of chromosomes during mitosis.

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Structure & Asssembly

Microtubules are composed of a single type of globular protein called tubulin, i.e., a dimer consisting of two polypeptides: α -tubulin and β -tubulin.

- Tubulin dimers polymerize to form microtubules, which generally consist - in general - of 13 linear protofilaments assembled around a hollow core.
- The protofilaments are arranged in parallel. Consequently, microtubules are polar structures with two distinct ends: a fast-growing plus end and a slow-growing minus end.
- This polarity is an important consideration in determining the direction of movement along microtubules

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Dynamic instability



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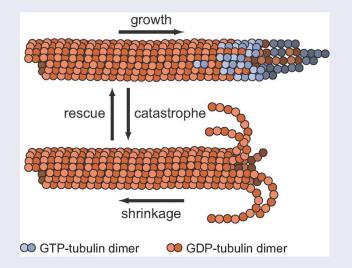
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Microtubules

Dynamic properties

Microtubules dynamic properties are tightly regulated by cellular microtubules associated proteins (MAPs) and are modulated in a variety of ways by many microtubule-targeted drugs.

- Some the most effective chemotherapeutics, such as the taxanes, are microtubule interfering drugs. Their efficacy has been demonstrated in the clinic for the treatment of a wide variety of human cancers, including breast, lung, ovarian, and prostate, as well as haematological malignancies and childhood cancers.
- Many studies suggest that microtubule dynamics are altered in cancer cell divisions.



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Protein-based cellular functions & EM fields

Being able to control protein-based cellular functions with an electromagnetic field could open an exciting spectrum of possibilities for advancing biotechnological processes. Besides, it paves the way for the development of new biomedical theranostic approaches to treat various diseases where specific proteins are known targets.

Theranostic

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Theranostics is a term derived from a combination of the words therapeutics and diagnostics. In this emerging field of medicine, drugs and/or techniques are uniquely combined to simultaneously or sequentially diagnose and treat medical conditions.



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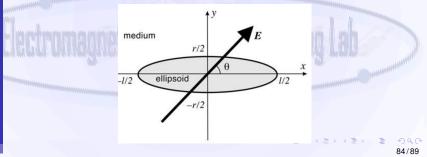
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Microtubules Electroorientation Its is a phenomenon that takes place when an electric field is acting on the microtubules

The microtubule can be modeled as an ellipsoid with rotational symmetry of radius r and length l.





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Microtubules Electroorientation When the ellipsoid is placed with its long axis at angle θ to the electric field E*, the electrostatic torque exerted on the ellipsoid is calculated by

$$\mathbf{T}_{\boldsymbol{e}} = 1/2\Re\left(\mathbf{p}^* \times \mathbf{E}^*\right) \tag{38}$$

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where p* is the dipole moment induced in the ellipsoid.
If we take the coordinate x in the direction of the longest axis of the ellipsoid and coordinate y perpendicular to it (both x and y are in the plane parallel to the direction of applied electric field), the orientation of the ellipsoid is caused by the torque about the z axis, which is perpendicular to the direction of electric field.



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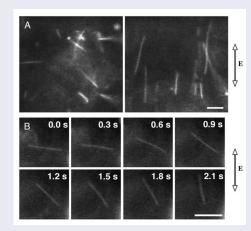
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- Cells
- Polarization

Biological tissues Debye model Dispersion in biological tissues non-Debye models Cole: Cole model

Microtubules Electroorientation

Electroorientation

- Orientation of microtubules is random in the absence of the electric field (A - Right).
- Upon application of the electric field, the microtubules are oriented parallel to the direction of the field line (A -Left).
- (B) Sequential images of a single microtubule after the onset of the electric field taken at intervals of 0.3µs.
- Arrows indicate the direction of the applied electric field.



Bioengineering

ERSLab F. Nunziata

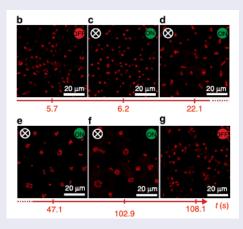
Why we do care Natural media Mixture

Propagation issues Average value Uniform plane wave Special media

Cells Polarization

Biological tissues Debye model Dispersion in biological tissues non-Debye models Cole-Cole model

Microtubules Electroorientation Simple ellipsoidal colloids can reversibly self-assemble into regular tubular structures when subjected to an alternating electric field





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